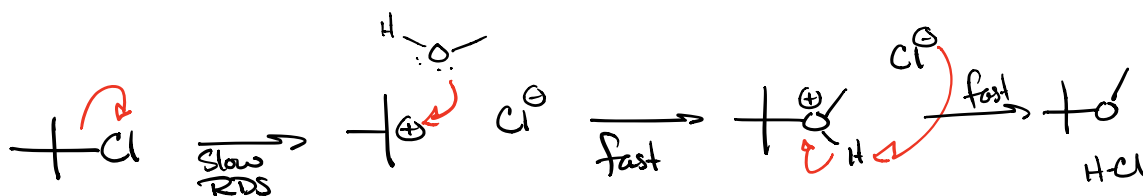


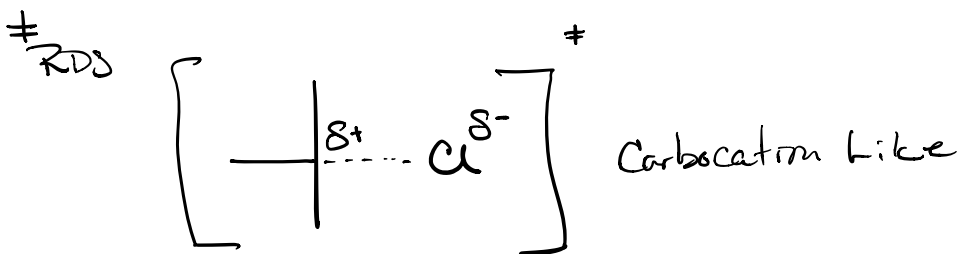
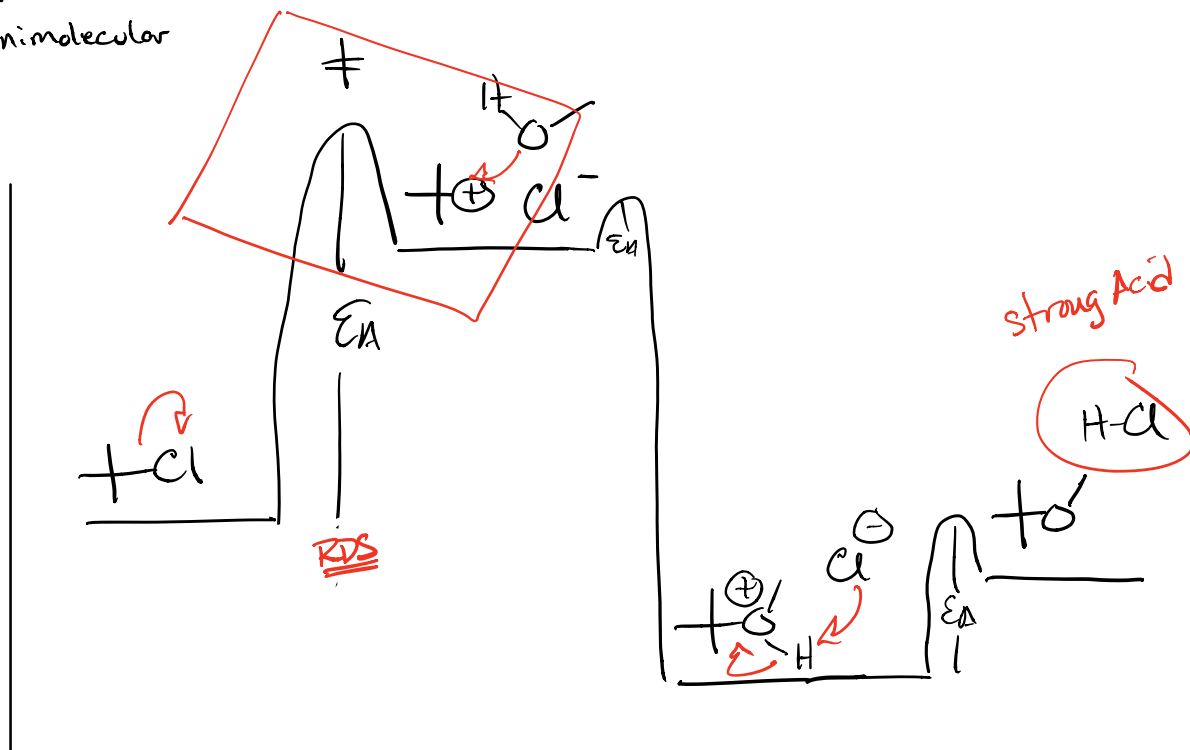
Solvolysis Replacement of LG by solvent



Dissociation & Carbocation formation
unimolecular

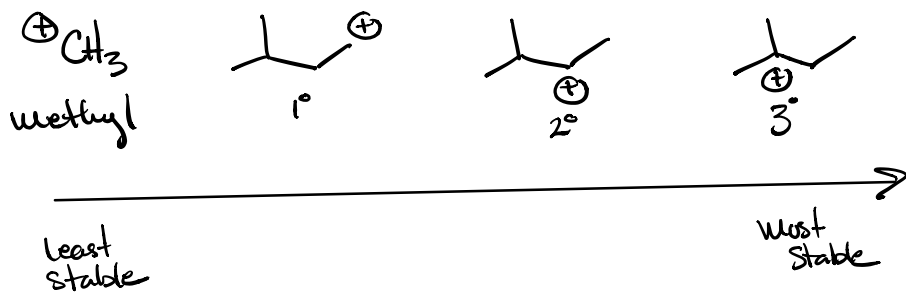
nucleophilic attack

Proton transfer

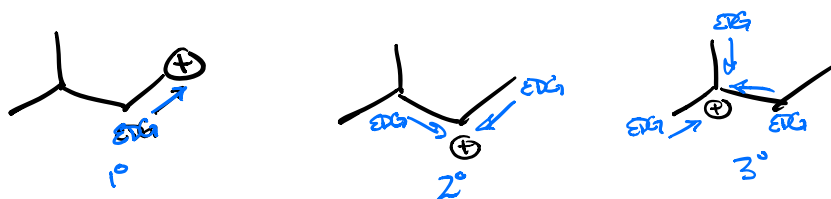


to understand $\text{S}_{\text{N}}1$ you must understand C^+

Carbocation Stability



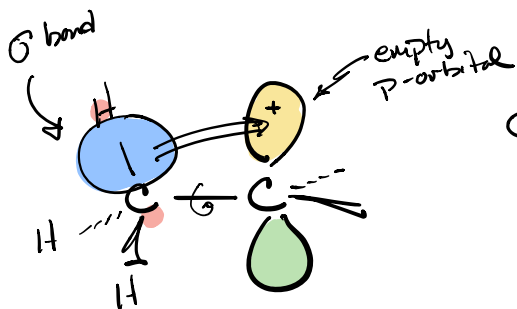
Why? ⇒ Remember that alkyl group EDG!



Why is an alkyl group EDG?

Induction

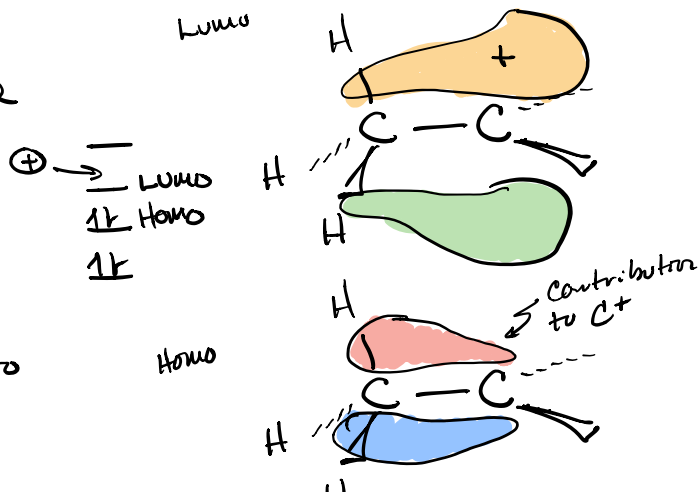
Hyperconjugation (Natural bonding theory) localized bonds

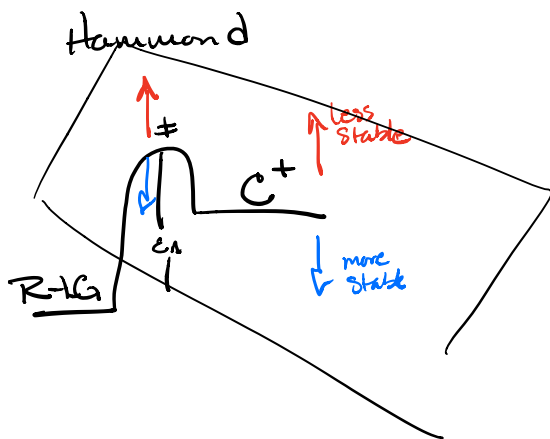
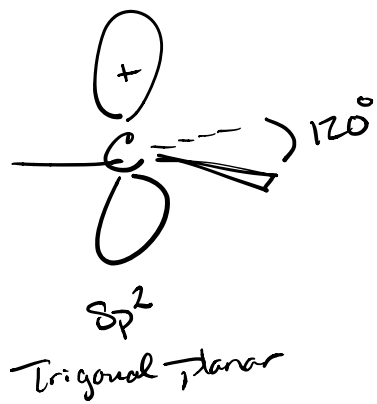
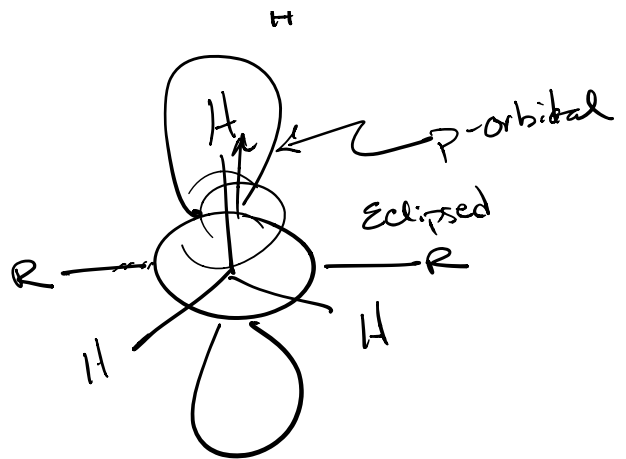
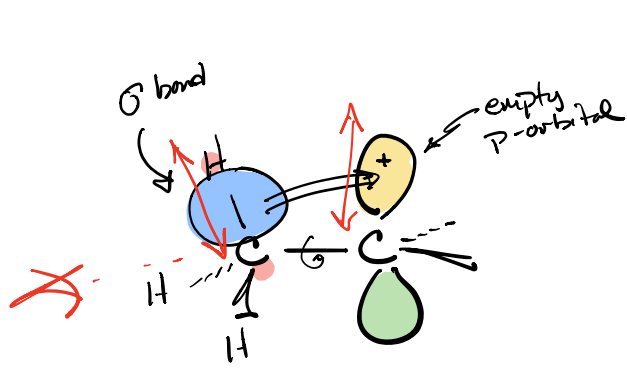


σ - dumps e⁻ density into empty p-orbital
Back bonding

Molecular Orbital Theory

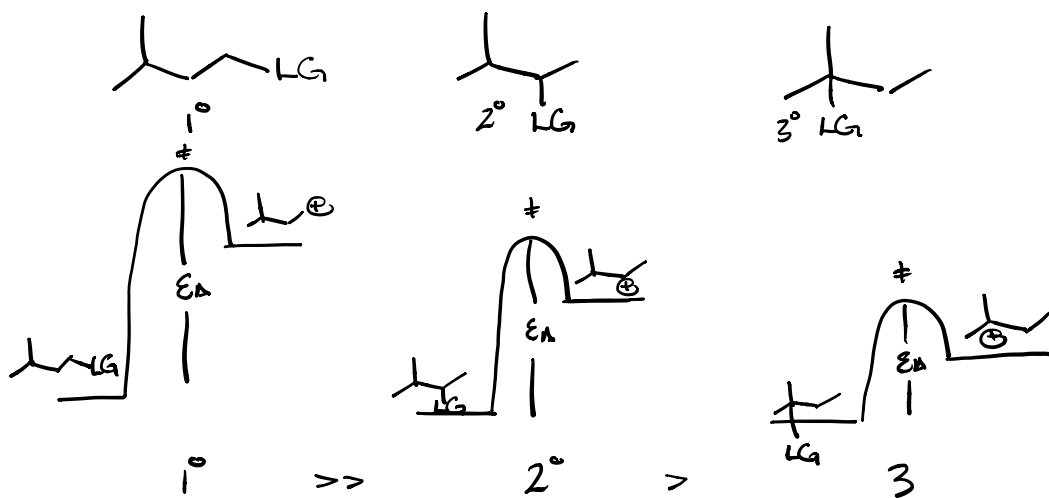
Molecular Orbitals





‡ Energy is linked to C⁺ energy

$$\ddagger \propto C^+$$



$$\text{Rate} = k [\text{R-LG}]$$

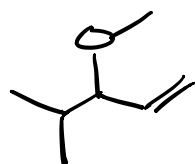
$$k \propto E_A$$

$$k = k_0 e^{-\frac{\Delta G^\ddagger}{RT}}$$

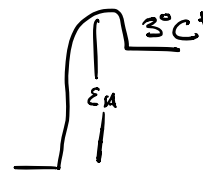
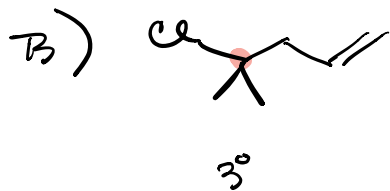
$$k = k_0 e^{-\frac{E_A}{RT}}$$

$$\left. \begin{array}{l} k = k_0 e^{-\frac{\Delta G^\ddagger}{RT}} \\ k = k_0 e^{-\frac{E_A}{RT}} \end{array} \right\} \Delta G^\ddagger = E_A$$

Which Run faster?



faster
Due to
Resonance



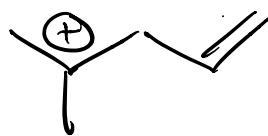
Don't look at Substrate ⇒ look at resulting C⁺



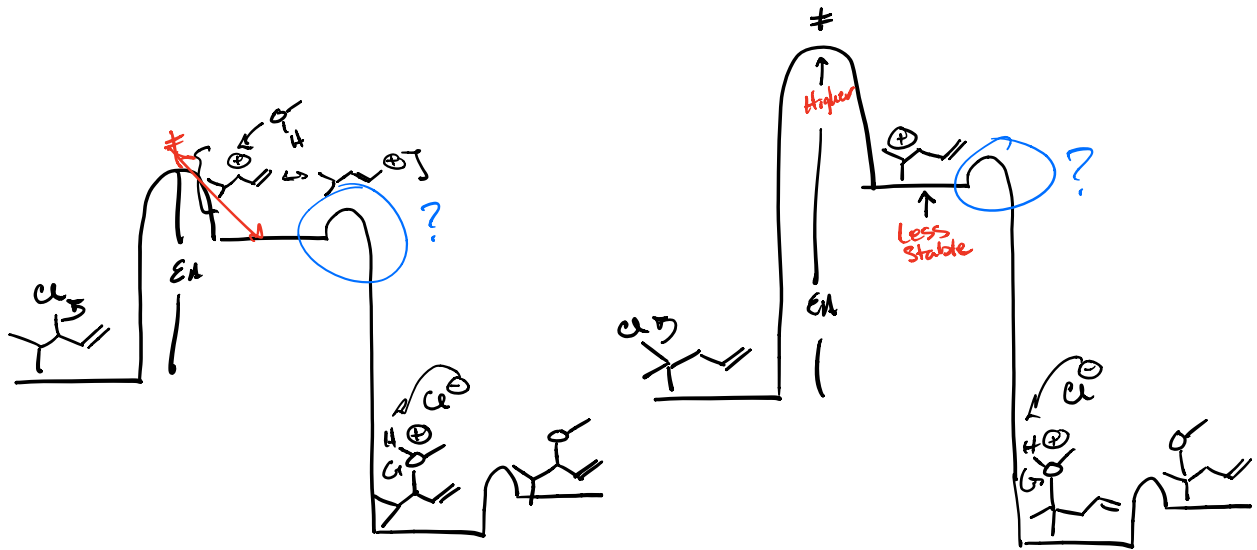
2° C⁺ ⇒ Allylic

faster

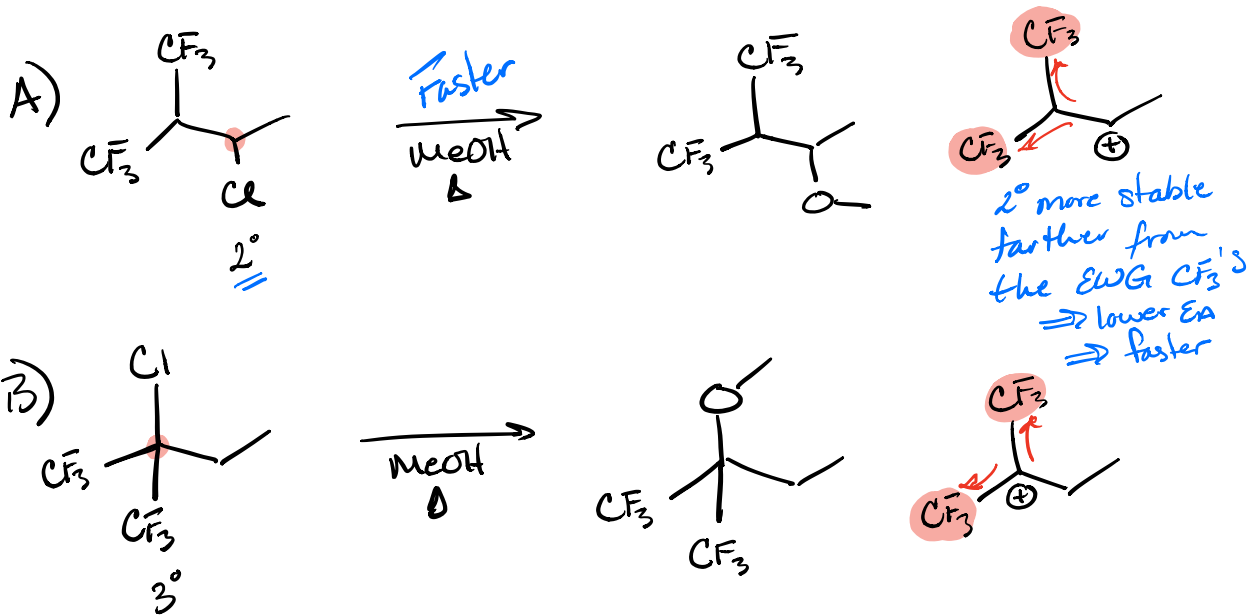
more stable C⁺ & EA
for formation

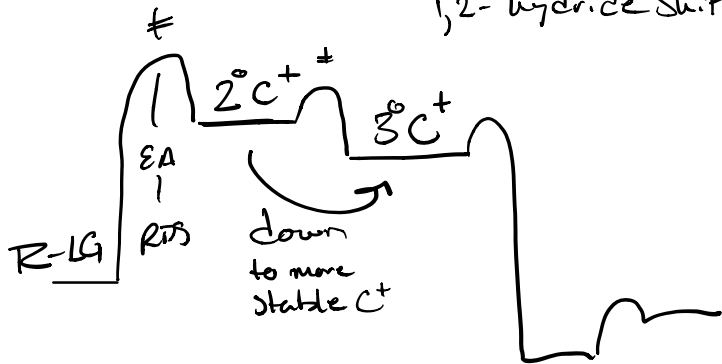
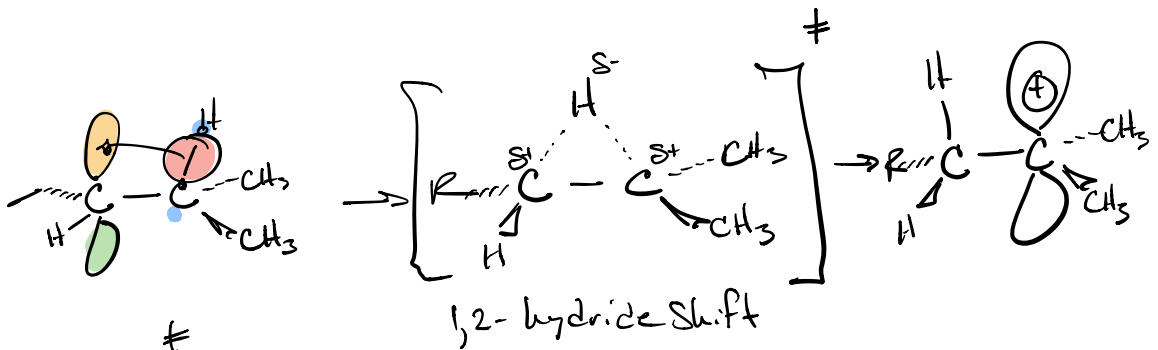
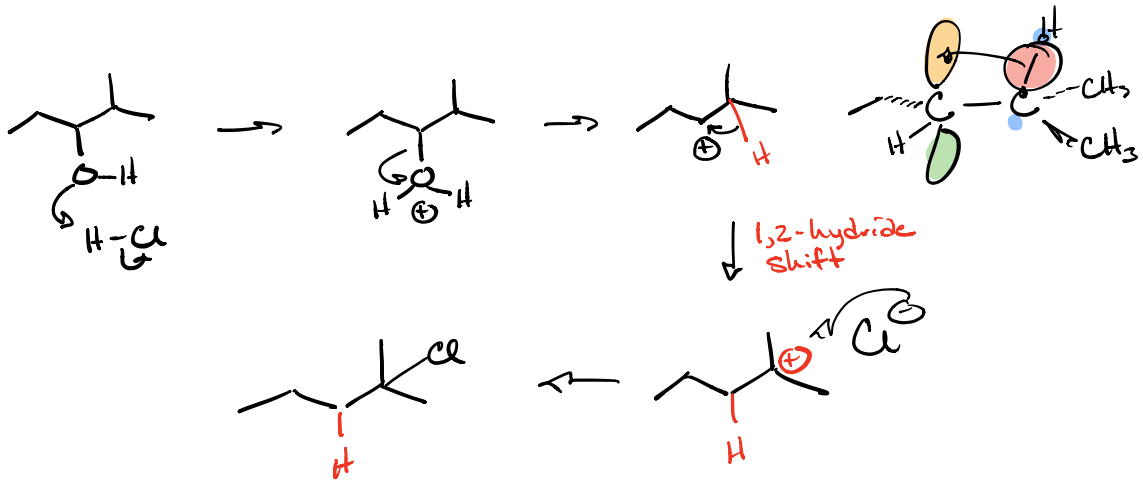
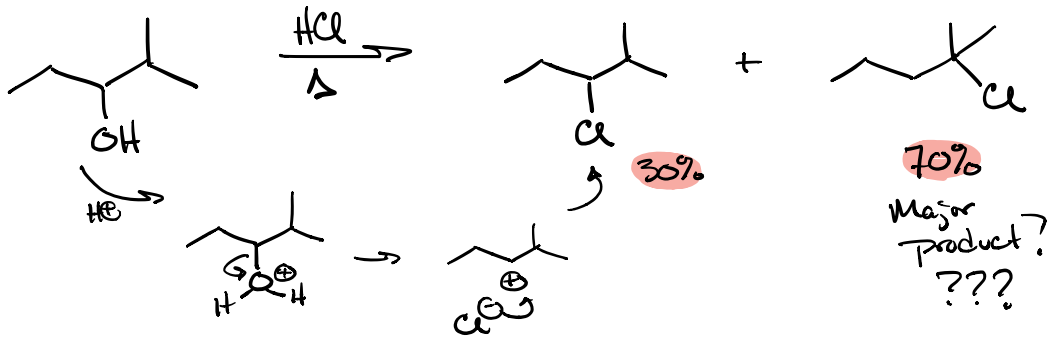


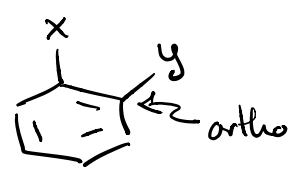
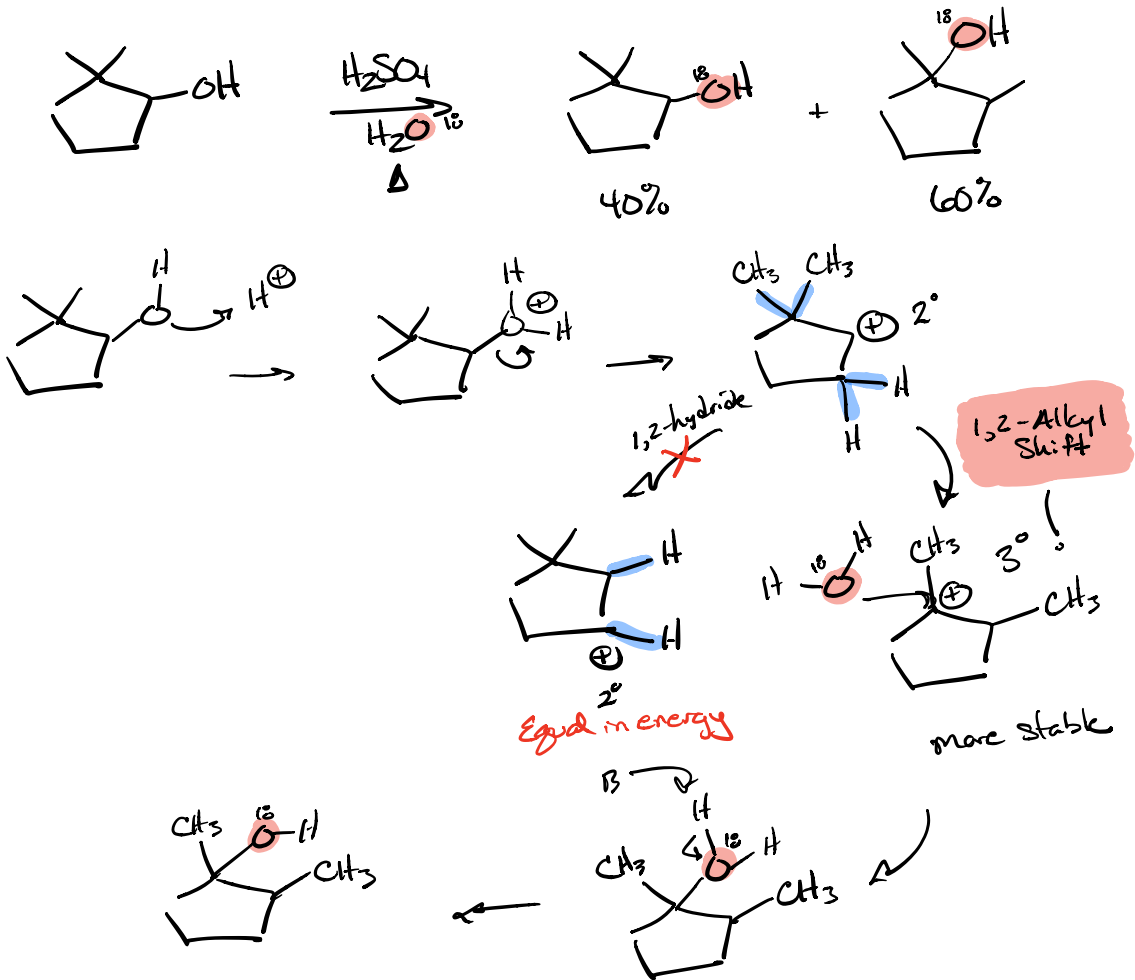
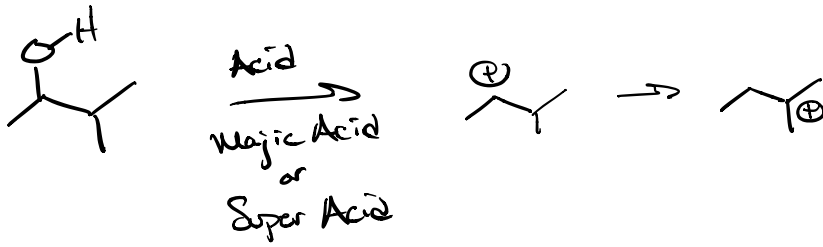
3° no resonance

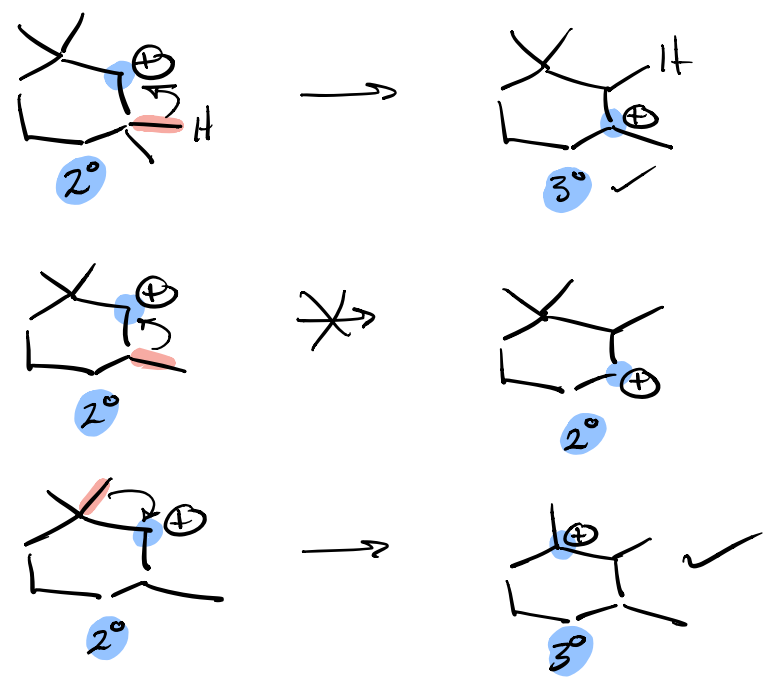
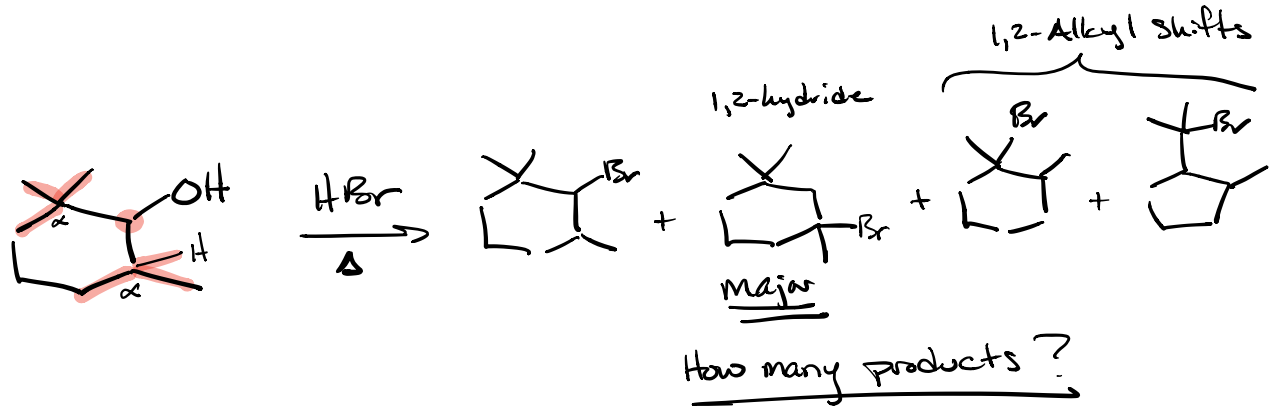
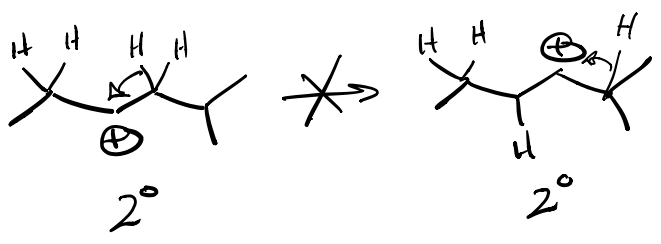
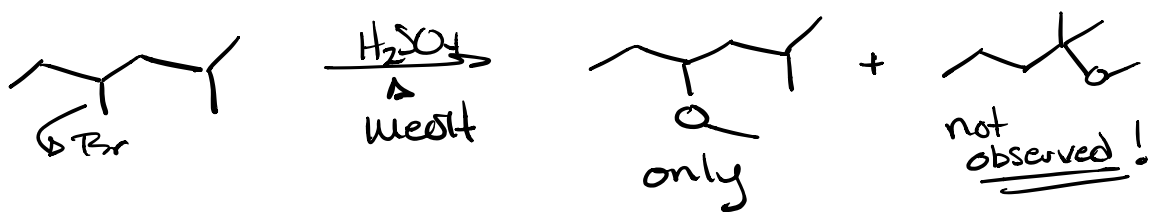


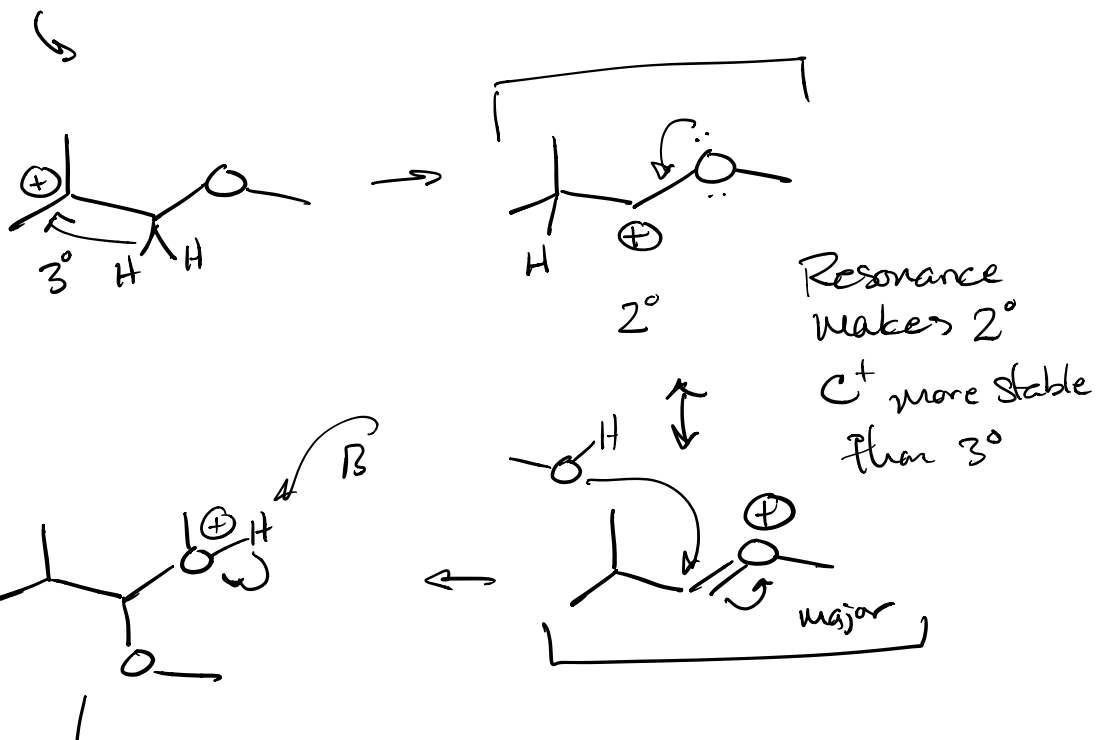
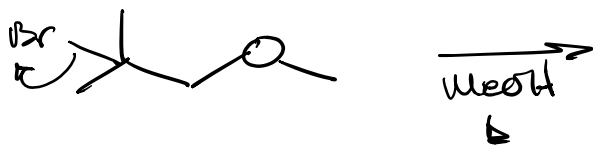
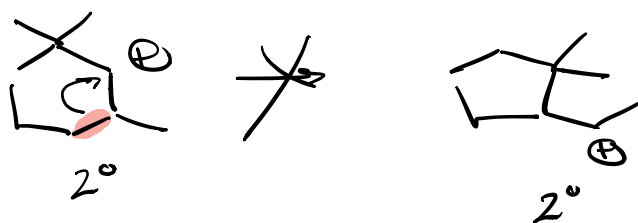
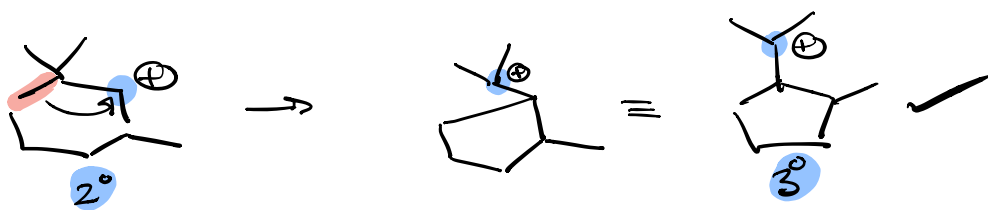
which R_{SN1} is faster?

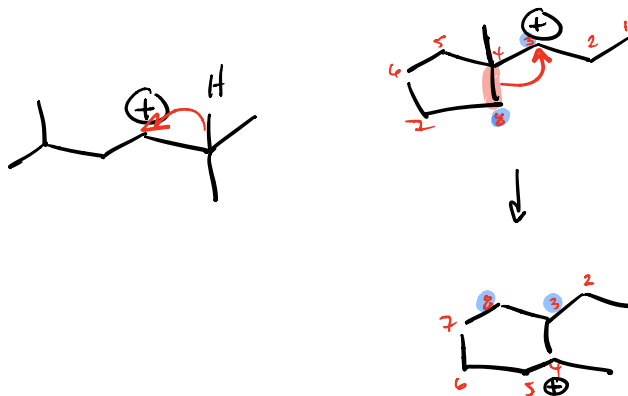
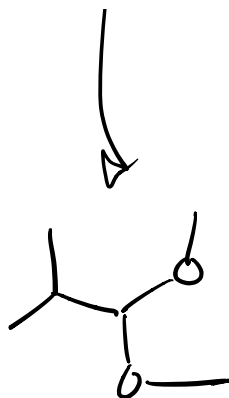












↑ EN/hybridization pKa 20-30

<u>Resonance</u>	10-15	←
Size	5-10	
<u>Induction</u>	0.2-1	←